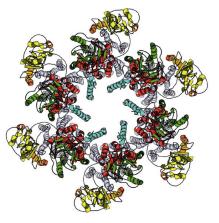
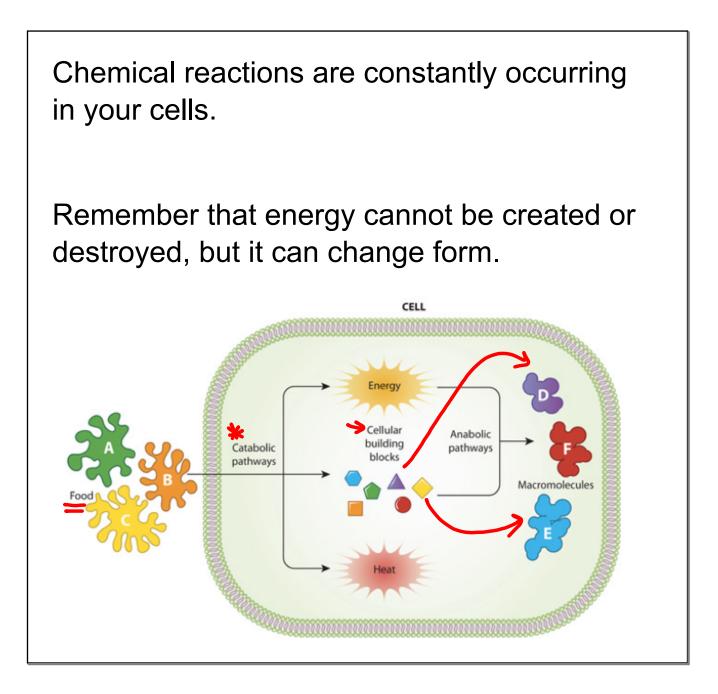
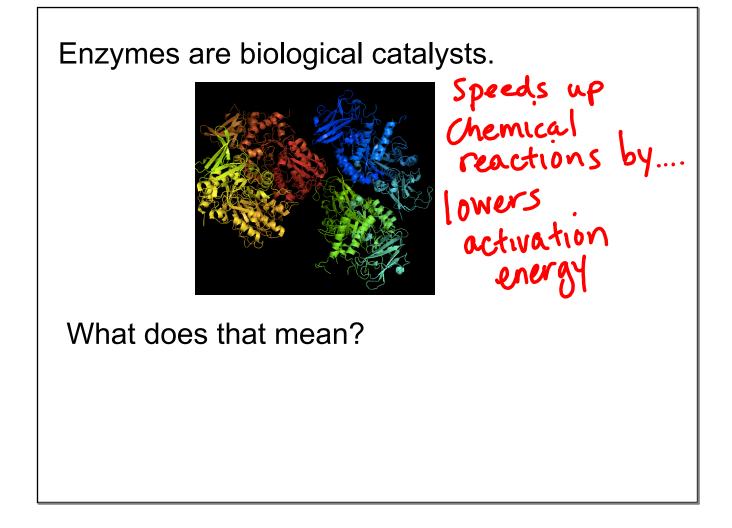
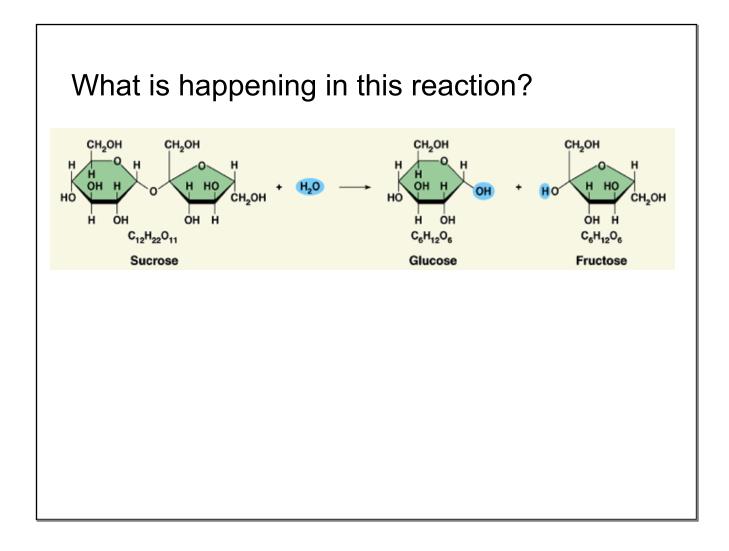
## Introduction to Enzymes



Nature Reviews | Molecular Cell Biology







## Some processes are **spontaneous** at room temperature. There is sufficient energy to overcome the energy barrier.

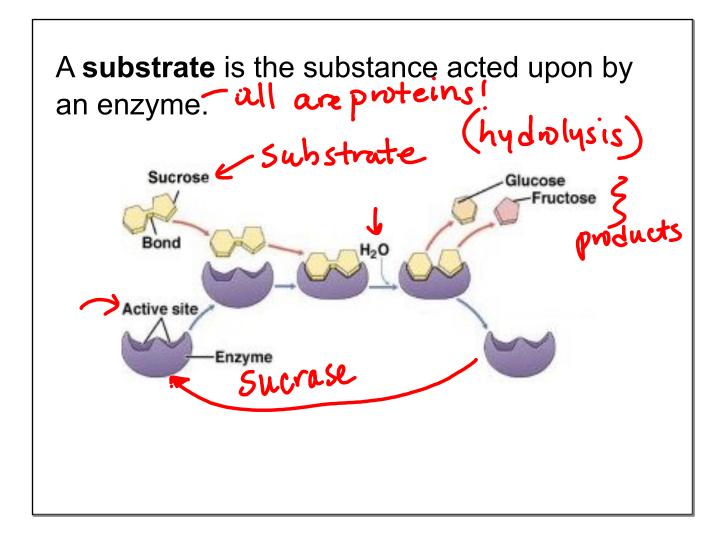


 $Zn(s) + Cu^{2+}(aq) \rightarrow Zn^{2+}(aq) + Cu(s)$ 

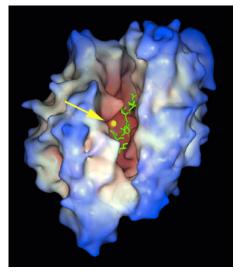
Other reactions require additional energy or help from a catalyst to occur at room temperature.

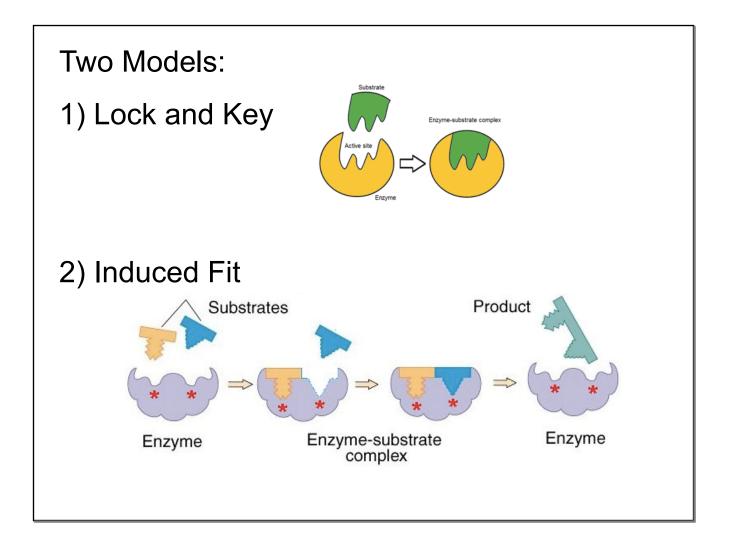
These reactions include the breakdown of proteins, DNA and other complex molecules.

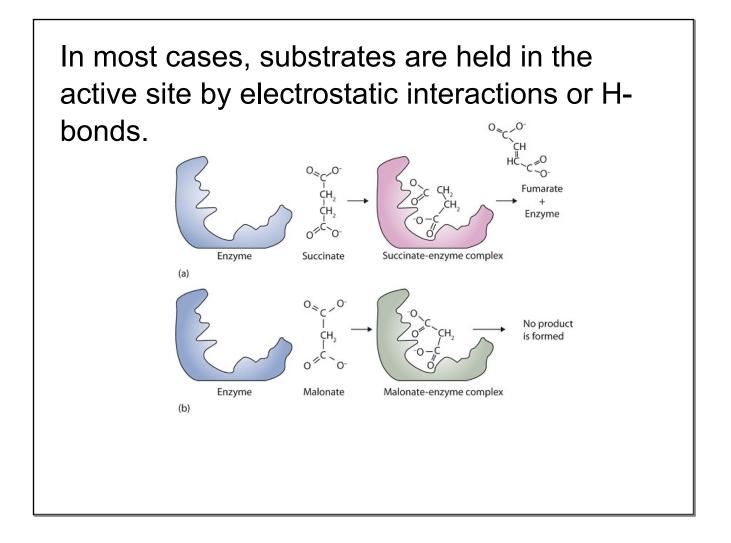
Why do we need these reactions? Why can't we just add heat to speed up the reactions?



The **active site** of an enzyme is a pocket or groove on the surface of the protein into which the substrate fits.







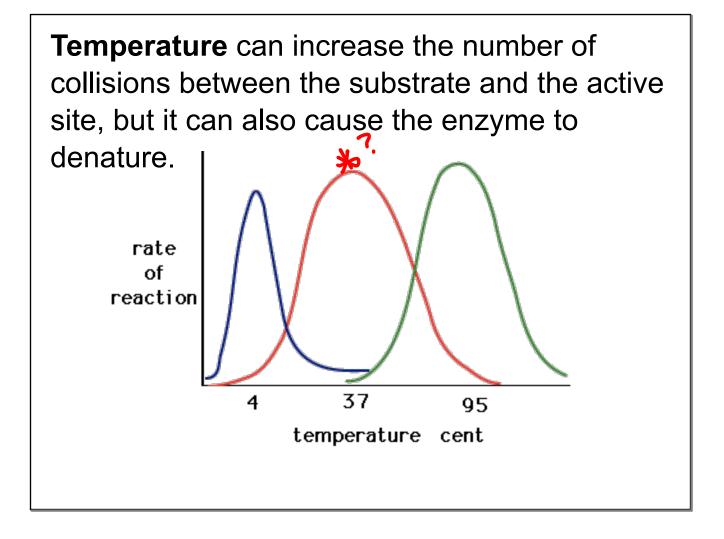
A single enzyme molecule can catalyze thousands of rounds of a reaction per second.

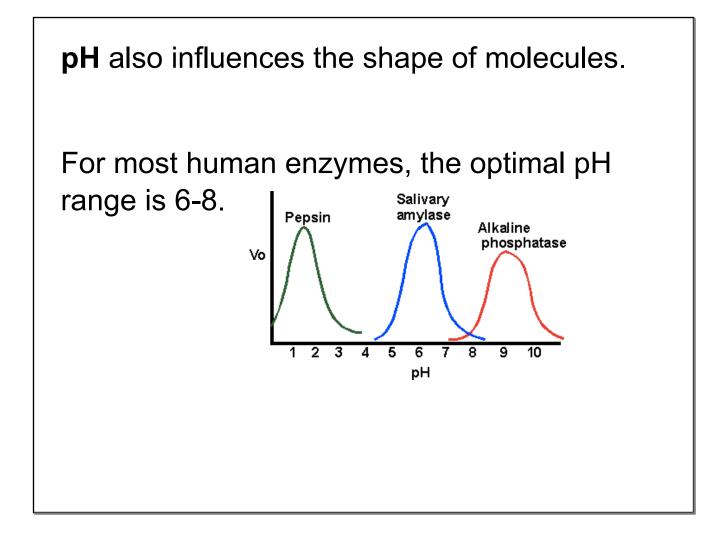
Enzymes are regenerated at the end of a reaction.

Most metabolic enzymes can catalyze both the forward and the reverse reaction. The direction depends on **equilibrium**. A cell's physical and chemical environment affects enzyme activity.

An enzyme needs specific higher order structures in order to function correctly.

What are some factors that affect enzyme activity?

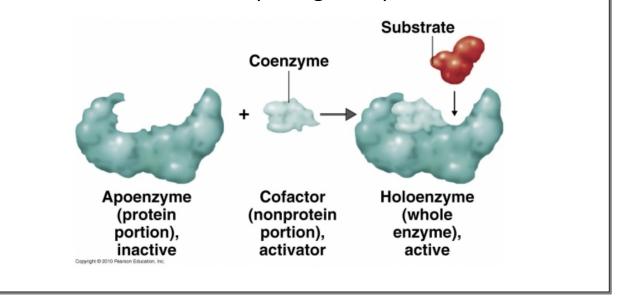




**Coenzymes** and **cofactors** can also affect enzyme activity.

Both work to change the active site conformation to improve binding to a substrate.

Coenzymes are vitamins (organic), whereas cofactors are metals (inorganic).



## **Inhibitors** (competitive or non-competitive) can slow down or stop enzyme activity.

Both types of inhibitors prevent the enzymesubstrate complex from forming.

